### UNIVERSITY OF PUERTO RICO AT ARECIBO PHYSICS/CHEMISTRY DEPARTMENT BACHELOR OF TECHNOLOGY IN INDUSTRIAL CHEMICAL PROCESSES

Course N°: QU	JIM 3025	Title of Course: Analytical Chemistry					
Credits: 4 Contact Hours: 3/weekly		<b>Open to</b> : Natural Science and Industrial Chemica Processes Technology students					
Pre- requisite:	QUIM 3002 / TEQU 3003	Textbook: Fundamental of Analytical Chemistry					
Co-requisite:	QUIM 3026	Author: Skoog, D.A., West, D.M., Holler, F.J., and Crouch, S.R.					
		Publisher: Brooks/Cole					
		<b>Publication Year: 2004</b>					

# **Other Supplemental Materials:**

- Analytical Chemistry for Technicians, by Kenkel, John. Boca Raton : Lewis Publishers, 2003.
- Quality and reliability in analytical chemistry, by Aboul-Enein, Hassan Y. Boca Raton : CRC Press, 2001.

Term: First Semester

Course Coordinator: Dr. Fernando L. Herrera

**Course Description:** Study of theory and methods for qualitative and quantitative analysis using gravimetric and volumetric methods, including potentiometric titration. Introduction to the spectrophotometric methods of analysis with emphasis on the ultraviolet-visible.

#### **Course Objectives:**

- Explain the principles that underlie the analysis methods commonly used such as volumetry, gravimetry, potentiometry, spectrophotometry, chromatography, analytical, radiochemistry, bioanalytical and environmental.
- Solve problems using data generated from analytical methods.
- Determine the statistical reliability of analysis results.
- Apply the knowledge gained to other areas such as pharmacy, biology and environmental sciences.

# **Relation of Course to Program Objectives:**

### **Relation of Course to Program Outcomes:**

	1		2	3		4				
	х									
1	2	3	4	5	6	7	8	9	10	11
X	X	X								

**Evaluation/Grade Reporting**: 4 partial examinations (15% each), assignments and quizzes (15%), and final exam (25%)

TopicsStrategies Time Distribution (hours)IntroductionLectures and Assignments (1.5)Statistical analysisLectures and Assignments (3)StoichiometryLectures and Assignments (1.5)	<b>Teaching/Learning</b>				
I opics Time Distribution (hours)   Introduction Lectures and Assignments (1.5)   Statistical analysis Lectures and Assignments (3)   Stoichiometry Lectures and Assignments (1.5)	Strategies				
(hours)IntroductionLectures and Assignments(1.5)(1.5)Statistical analysisLectures and Assignments(3)StoichiometryLectures and Assignments(1.5)	Time Distribution				
IntroductionLectures and Assignments (1.5)Statistical analysisLectures and Assignments (3)StoichiometryLectures and Assignments (1.5)					
(1.5) Statistical analysis Lectures and Assignments (3) Stoichiometry Lectures and Assignments (1.5)					
Statistical analysisLectures and Assignments(3)StoichiometryLectures and Assignments(15)					
(3) Stoichiometry Lectures and Assignments (1.5)					
Stoichiometry Lectures and Assignments					
(15)					
(1.5)					
General concepts of balance Lectures and Assignments					
(4.5)					
Gravimetric analysis Lectures and Assignments					
(2.5)					
Acid-base Lectures and Assignments					
(15)					
Titration curves of acid-base Lectures and Assignments					
(3.5)					
Reactions and degrees complexometric Lectures and Assignments					
(3)					
Reactions and degrees of precipitation Lectures and Assignments					
(1.5)					
Basic concepts in instrumental analysis Lectures and Assignments					
Fundamentals of electrochemistry Lectures and Assignments					
(1.5)					
Potentiometry Lectures and Assignments					
(4.5)					
Molecular absorption Lectures and Assignments					
spectrophotometry (4)					
Chromatographic methods Lectures, and Assignments	•				
(3)					
Total 45					